

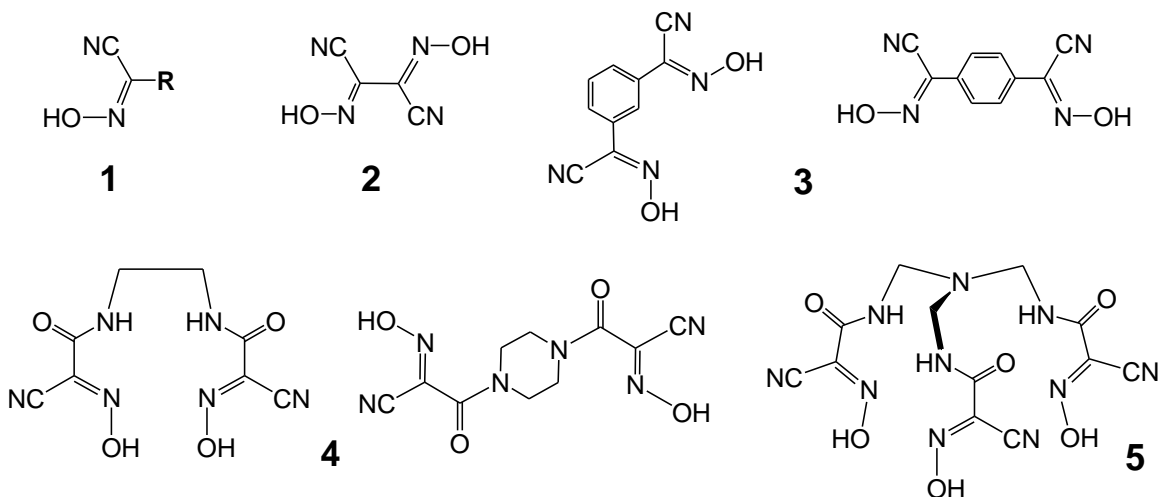
CHEMISTRY AND APPLICATIONS OF CYANOXIMES AND THEIR METAL COMPLEXES

by

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During the last three decades research interests were dedicated to the new subclass of organic ligands – cyanoximes – that have general formula NC-C(=NOH)-R where R is an electron-withdrawing group. Presence of CN-group makes cyanoximes ~10,000 more acidic and better ligands than other known oximes. With 38 different R groups the most abundant is the family of *mono*-cyanoximes **1** (below), followed by *bis*-cyanoximes **2-4** that include aromatic and aliphatic spacers, and lately *tris*-cyanoxime **5** - a tripod - was obtained and characterized.



These simple low molecular weight organic molecules represent series of new excellent ampolydentate ligands for coordination chemistry: new types of *molecular Legos*. Both un-complexed ligands, their Na^+ and K^+ salts and other metal complexes show a large spectrum of biological activity from growth regulation in plants to significant *in vitro* and *in vivo* cytotoxicity against human cancers. Currently 44 cyanoximes are known, and there were more than two hundreds cyanoxime complexes synthesized and studied using a variety of different spectroscopic methods and X-ray analysis. Methods of syntheses, stereochemistry of cyanoximes ligands, their structures and properties, as well as the most interesting coordination compounds are reviewed in this presentation. A broad spectrum of practical applications of both cyanoximes, and their metal complexes is outlined in this lecture as well.

Some of the most representative citations:

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